PHYSICAL SCIENCES

PAPER 1

| ~ | | | |
|---|--|-------------------------------|------------|
| • | Drawing a labelled free body diagram Stating one of the Newton's law of motion About 2 calculations, at least 2 marks for the formulae + 2 marks for substantial contents. | (3) (2) titution [9] | (4) |
| TIPS | | | |
| * | Emphasise the use of simultaneous equations | | |
| QUES | ΓΙΟΝ 3: VERTICAL PROJECTILE MOTION | | |
| • | 2/3 calculations, formulae and substitution Using information from the graph/ drawing a graph | (4) (2) [6] | |
| TIPS | | | |
| | Sign conventions Differentiate between Δt and t Show frame of reference Emphasise drawing and interpretation of graphs | | |
| QUES | ΓΙΟΝ 4: MOMENTUM | | |
| • | Defining a term/stating a law/ principle (Impulse, elastic/non-elastic coprinciple of linear momentum) 2/3 calculations | llision, | (2) (4) |
| TIPS | | | |
| Be able to identify the initial and final velocity Emphasise subscripts and understand what they mean especially net in F_{net}Δt=Δp | | | |
| QUES | ΓΙΟΝ 5: WORK, ENERGY & POWER | | |
| • | Defining a term/ stating a law/principle (Conservative/non-conservative) Conservation of mechanical energy) 2/3 calculations | e force, (2) (4) [6] | |

TIPS

| * | Label free body diagrams with arrows starting from the point Correct choice of formulae Differentiate between W_{net} & F_{net} Differentiate between $W = F\Delta x Cos$ | |
|----------|---|-------------------|
| QUES | TION 6: DOPPLER EFFECT | |
| • | Defining a term (Doppler effect) Calculation | (2) (3) |
| | | [5] |
| TIPS | | |
| | Write the correct formula from the data sheet and know each symbol Mathematical manipulation | |
| QUES | TION 7: ELECTROSTATICS (COULOMBS LAW) | |
| • | Defining a term/ stating a law (Coulombs law, electric field) Direction of the electric field Calculation | (2) (1) (3) |
| | | [6] |
| TIPS | | |
| | Integrate with Newton's laws concepts Labelling of forces | |
| QUES | TION 8: ELECTROSTATICS (ELECTRIC FIELDS) | |
| • | Defining a term/ stating a law (Ohm's law) Calculation Interpreting a graph/using information from the graph | (2) (3) [5] |
| TIPS | | |
| * | Choice of formulae Direction of the net field | |

Emphasise the vector nature of electric fields

QUESTION 9: ELECTRIC CIRCUITS

| • Calcul | lation | (3) |
|---|---|------------|
| | Using information from the graph | (3) |
| | <u> </u> | [3] |
| TIPS | | |
| MatheUse ofApplieDifferent | e of formulae ematical manipulation f correct units cation of Ohm's law entiate between EMF and potential difference e experiments stated in CAPS | |
| QUESTION 1 | 0: ELECTRODYNAMICS | |
| • 2/3 ca | lculations | (4) [4] |
| TIPS | | |
| EmphBe abl | the difference between AC & DC circuits asise the use of subscripts e to interpret Faraday's law | |
| QUESTION I | 1: PHOTOELCTRIC EFFECT | |
| | ing a term (Work function, threshold frequency, cut-off frequency) | (2) |
| - | olculations ng use of the graph | (4) |
| | | [6] |
| TIPS | | |
| | the gradient to hc e to use the graph to determine the threshold frequency and Planck' | s constant |
| | TOTAL | = 56/150 |
| | | 37% |

NB: NOT INCLUSIVE OF MULTIPLE CHOICE QUESTIONS. 4 marks from MCQ will add up to $60/150\text{=}\,40\%$

PHYSICAL SCIENCES P 2

| QUES | 11ON 2, 3 & 4: INORGANIC MOLECULES | | | |
|------|---|----------|--------|------------|
| • | Functional groups, homologous series IUPAC naming, formulae of compounds (2) | | | |
| • | Structural formulae | | | (2) |
| • | Definitions of terms. hydrocarbon, isomer, saturated, unsaturated, functional group et (2) | | | roup etc |
| • | Positional, functional and structural isomers | | | (2) |
| • | Reaction conditions and types of reactions | | | (2) |
| • | Dependant and independent variables | | | (2) |
| • | Polymerisation, addition and condensation polymers (differentiat | e) | | (2) |
| • | Intermolecular forces at least | , | | (2) |
| • | Reactants and products of combustion and balancing the equation | S | | (3) |
| | 91 | | ± | [20] |
| WOUI | HE FOLLOWING DOES NOT NECESARILY MEAN THAT ALL TH LD BE ASKED TION 5: RATE OF REACTIONS | IE QUI | ESTIC | ONS |
| ~ | | | | |
| • | Experiment investigating factors that influence the reaction rate Interpreting the graph/table | | | (2) (2) |
| • | Balancing equations, stating if they are exothermic or endothermic | <u>;</u> | | (3) |
| • | Calculation at least | | | (2) |
| | | | ± | [9] |
| TIPS | | | | |
| * | Revise stoichiometric calculations | | | |
| * | Emphasise skills for interpreting the graphs | | | |
| * | The use of the correct formulae | | | |
| | | | | |
| QUES | TION 6: CHEMICAL EQUILIBRIUM | | | |
| • | Stating Le Chateliers principle | | | (2) |
| • | Defining terms, reaction rate, reversible reaction, chemical/dynam | uic eani | libriu | • • |
| • | K_c calculation, at least 4 marks from the K_c expression, 2 from the t | - | | • • |
| | Substitution | | 111 | (4) |
| | | ± | | [8] |
| | | | | |
| | | | | |

TIPS

- ❖ Identify the disturbance
- ❖ State which reaction will be favoured when opposing the disturbance
- ❖ The effect on what happens to the number of moles of the products etc
- \diamond Different methods of calculating K_c
- Drawing of Boltzmann distribution curve

QUESTION 7: ACIDS AND BASES

| • | Defining terms, acid and bases according to Lowry Bronsted/Arhenius, Ampholyte | | lyte |
|---|--|---|------|
| | Mono & diprotic acids | | (2) |
| • | Conjugate acids/bases | | (2) |
| • | Know why acids are weak/strong/monoprotic and diprotic | | (1) |
| • | Know all the apparatuses for titration | | (2) |
| • | Calculation, at least 3 marks | | (3) |
| | | ± | [10] |

TIPS

❖ Emphasise the use of correct formulae and substitution

QUESTION 8: ELECTROCHEMICAL CELLS

| • | Functions of the salt bridge | | (2) |
|---|--|----------|------|
| • | Standard conditions under which a cell functions | | (2) |
| • | Cell notations | at least | (2) |
| • | Oxidising/reducing agents(name/formula) | at least | (1) |
| • | Energy conversions | | (1) |
| • | Calculating the EMF at least 3 marks | | (3) |
| | | ± | [11] |

QUESTION 9: ELECTROLYTIC CELL

| • | Defining terms, electrolyte, electrolysis | (2) |
|---|---|-----|
| • | Writing ½ reactions | (2) |
| • | Oxidising/reducing agents(name/formula) | (2) |
| • | Type of electrode cathode/anode | (1) |
| | • • | [7] |

TIPS Q 8 & 9

- ❖ Emphasise the writing of ½ reactions, the use of double arrows
- ❖ Master the use of the table of standard reduction potentials

QUESTION 10: FERTILIZERS

| • Table | |
|---|-----|
| Writing a balanced equation | (3) |
| Industrial preparations | (1) |
| Name/formula of the catalysts, acids, fertilizers | (1) |
| • Calculations | (2) |
| | [7] |

TIPS

- Know all the industrial processes
- ❖ Be able to use ratios in calculations
- Emphasise conversions